

## *rac*-2-[2-(4-Fluorophenyl)-2-oxo-1-phenylethyl]-4-methyl-3-oxo-*N*-phenylpentanamide

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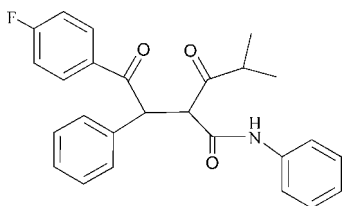
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Key indicators: single-crystal X-ray study;  $T = 173$  K; mean  $\sigma(\text{C}-\text{C}) = 0.006$  Å;  $R$  factor = 0.067;  $wR$  factor = 0.227; data-to-parameter ratio = 13.8.

The title compound,  $\text{C}_{26}\text{H}_{24}\text{FNO}_3$ , is a critical intermediate of a selective and competitive inhibitor of the enzyme 3-hydroxy-3-methylglutaryl-coenzyme A (HMG-CoA) reductase. Intermolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonding generates a chain along **[give direction]** that is the dominant interaction in the crystal packing. Intermolecular  $\text{C}-\text{H}\cdots\text{O}$  interactions are also observed.

### Related literature

For related structures, see: Baumann *et al.* (1992). For the title compound as an intermediate in the preparation of the HMG-CoA reductase inhibitor atorvastatin, see: Roth *et al.* (1991); Wang *et al.* (2007).



### Experimental

#### Crystal data

$\text{C}_{26}\text{H}_{24}\text{FNO}_3$   
 $M_r = 417.46$

Monoclinic,  $P2_1/n$   
 $a = 14.1694$  (14) Å

$b = 9.8307$  (9) Å  
 $c = 16.6367$  (16) Å  
 $\beta = 99.651$  (2)°  
 $V = 2284.6$  (4) Å<sup>3</sup>  
 $Z = 4$

Mo  $K\alpha$  radiation  
 $\mu = 0.08$  mm<sup>-1</sup>  
 $T = 173$  K  
 $0.15 \times 0.10 \times 0.05$  mm

#### Data collection

Bruker SMART CCD area-detector diffractometer  
Absorption correction: multi-scan (SADABS; Bruker, 2005)  
 $T_{\min} = 0.990$ ,  $T_{\max} = 0.996$

10555 measured reflections  
3792 independent reflections  
2493 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.027$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.067$   
 $wR(F^2) = 0.227$   
 $S = 1.06$   
3792 reflections  
275 parameters

1 restraint  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.78$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.68$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{O1}^i$	0.88	2.10	2.959 (2)	178 (4)
$\text{C1}-\text{H1A}\cdots\text{O3}^{ii}$	0.95	2.43	3.371 (5)	169

Symmetry codes: (i)  $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{3}{2}$ ; (ii)  $x - \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$ .

Data collection: SMART (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: KP2271).

### References

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**supplementary materials**

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## ***rac*-2-[2-(4-Fluorophenyl)-2-oxo-1-phenylethyl]-4-methyl-3-oxo-*N*-phenylpentanamide**

**F. Zhou and J. Huang**

### **Comment**

The title compound, C<sub>26</sub>H<sub>24</sub>FNO<sub>3</sub>(I), (Fig. 1) is of value as a pharmaceutical intermediate, particularly an intermediate of an HMG-CoA reductase inhibitor, atorvastatin (Roth *et al.*, 1991; Wang *et al.*, 2007). Though this compound reveals an opposite chirality at C8 and C9, up to now, the absolute configuration has not been reported so far, enantiomers separation or stereospecific synthesis of enantiomer of the title compound. A racemic mixture is always directly used to prepare HMG-CoA reductase inhibitor. Using *N*-ethyl thiazolium bromide as the catalyst, reacting 4-fluorobenzaldehyde with benzylidene isobutyrylacetamide affords I, as a white solid (Baumann *et al.*, 1992). Suitable crystals of I for single-crystal X-ray analysis were obtained by a vapor diffusion method. In the crystal structure intermolecular hydrogen bond N—H...O is a dominant interaction forming a chain (Table 1).

### **Experimental**

2-Benzylidene isobutyrylacetamide (2.93 g, 10.0 mmol), *N*-ethyl thiazolium bromide (0.290 g, 1.50 mmol), 4-fluorobenzaldehyde (1.36 g, 11.0 mmol), and triethylamine (8.08 g, 8.00 mmol) were mixed and heated to 338 K. The reaction mixture was stirred for 24 h maintaining the temperature at 338 K. After 2-propanol (5 ml) and water (5 ml) were added, a white precipitate was formed. The precipitate was filtered, washed with 2-propanol, and dried to afford the title compound (2.09 g, 50.2%) as a white solid. Colourless crystals were obtained by vapor diffusion of pentane into an acetone solution over a period of 5 d. <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN, 22 °C): δ 8.48 (1 H, s), 8.10 (2 H, m), 7.37 (2 H, m), 7.25 (6 H, m), 7.18 (3 H, m), 7.07 (1 H, m), 5.43 (1 H, d, *J* = 13.5 Hz), 4.74 (1 H, d, *J* = 13.5 Hz), 2.96 (1 H, m), 1.23 (3 H, d, *J* = 8.5 Hz), 1.02 (3 H, d, *J* = 8.0 Hz). <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>CN, 22 °C): δ 209.1, 197.5, 167.5, 165.5, 138.7, 136.2, 133.4, 132.5, 129.9, 129.8, 129.7, 128.8, 125.5, 120.9, 116.6, 64.2, 53.6, 40.7, 19.3, 18.2. ESI-MS: 440.2 [*M* + Na]<sup>+</sup>.

### **Refinement**

All H atoms were placed in calculated positions and refined as riding, with C—H = 0.95–1.00 Å, and N—H = 0.88 Å, and  $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5 U_{\text{eq}}(\text{C}, \text{N})$ .

### **Figures**

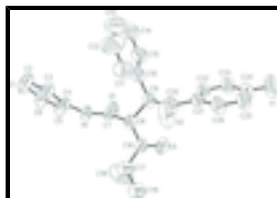


Fig. 1. The molecular structure with atom labels and 45% probability displacement ellipsoids for non-H atoms.

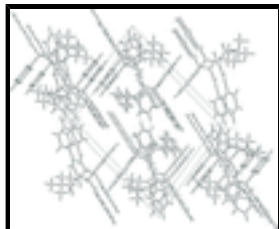


Fig. 2. The packing diagram of molecular, viewed down the *b* axis, with the N—H···O interactions shown as dashed lines.

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### Crystal data

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Monoclinic,  $P2_1/n$

Hall symbol: -P 2yn

$a = 14.1694$  (14) Å

$b = 9.8307$  (9) Å

$c = 16.6367$  (16) Å

$\beta = 99.651$  (2)°

$V = 2284.6$  (4) Å<sup>3</sup>

$Z = 4$

$F(000) = 880$

$D_x = 1.214$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 2549 reflections

$\theta = 2.4$ – $21.8^\circ$

$\mu = 0.08$  mm<sup>-1</sup>

$T = 173$  K

Prism, colourless

$0.15 \times 0.10 \times 0.05$  mm

### Data collection

Bruker SMART CCD area-detector diffractometer

Radiation source: fine-focus sealed tube graphite

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan (*SADABS*; Bruker, 2005)

$T_{\min} = 0.990$ ,  $T_{\max} = 0.996$

10555 measured reflections

3792 independent reflections

2493 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.027$

$\theta_{\max} = 24.5^\circ$ ,  $\theta_{\min} = 1.8^\circ$

$h = -16 \rightarrow 15$

$k = -11 \rightarrow 11$

$l = -14 \rightarrow 19$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.067$

$wR(F^2) = 0.227$

$S = 1.06$

3792 reflections

275 parameters

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1129P)^2 + 1.1017P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.78$  e Å<sup>-3</sup>

1 restraint

$$\Delta\rho_{\min} = -0.68 \text{ e } \text{\AA}^{-3}$$

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.0713 (3)	0.2839 (5)	0.9806 (3)	0.0948 (13)
H1A	0.0361	0.2691	1.0236	0.114*
C2	0.0430 (3)	0.3814 (5)	0.9239 (3)	0.0996 (14)
H2A	-0.0107	0.4368	0.9286	0.120*
C3	0.0911 (2)	0.4006 (4)	0.8599 (2)	0.0769 (10)
H3A	0.0698	0.4674	0.8196	0.092*
C4	0.1710 (2)	0.3222 (3)	0.85420 (17)	0.0521 (7)
C5	0.2012 (3)	0.2269 (3)	0.9127 (2)	0.0763 (10)
H5A	0.2569	0.1745	0.9100	0.092*
C6	0.1503 (4)	0.2071 (4)	0.9758 (2)	0.0958 (14)
H6A	0.1705	0.1398	1.0159	0.115*
C7	0.2586 (2)	0.2567 (3)	0.74381 (17)	0.0499 (7)
C8	0.2898 (2)	0.3116 (3)	0.66692 (17)	0.0546 (7)
H8A	0.2728	0.4103	0.6617	0.066*
C9	0.2357 (2)	0.2362 (3)	0.59207 (18)	0.0617 (8)
H9A	0.2563	0.1388	0.5948	0.074*
C10	0.1280 (3)	0.2419 (4)	0.5882 (2)	0.0743 (10)
C11	0.0794 (4)	0.3651 (5)	0.5882 (3)	0.1127 (15)
H11A	0.1140	0.4482	0.5908	0.135*
C12	-0.0191 (4)	0.3675 (8)	0.5845 (4)	0.142 (2)
H12A	-0.0511	0.4524	0.5845	0.170*
C13	-0.0706 (5)	0.2506 (10)	0.5810 (4)	0.146 (2)
H13A	-0.1381	0.2531	0.5782	0.175*
C14	-0.0231 (5)	0.1283 (8)	0.5815 (3)	0.131 (2)
H14A	-0.0580	0.0456	0.5793	0.158*
C15	0.0746 (3)	0.1249 (5)	0.5851 (2)	0.0931 (12)
H15A	0.1059	0.0394	0.5855	0.112*
C16	0.3973 (2)	0.2978 (3)	0.6729 (2)	0.0695 (9)
C17	0.4603 (3)	0.3732 (5)	0.7404 (3)	0.1074 (10)
H17A	0.4183	0.4299	0.7697	0.129*
C18	0.5129 (5)	0.2730 (6)	0.8006 (4)	0.168 (3)

## supplementary materials

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H18A	0.5533	0.3225	0.8447	0.253*
H18B	0.5531	0.2142	0.7728	0.253*
H18C	0.4665	0.2173	0.8234	0.253*
C19	0.5271 (3)	0.4666 (5)	0.7060 (3)	0.1074 (10)
H19A	0.5669	0.5158	0.7505	0.161*
H19B	0.4898	0.5319	0.6690	0.161*
H19C	0.5681	0.4133	0.6759	0.161*
C20	0.2594 (3)	0.2989 (3)	0.5144 (2)	0.0752 (10)
C21	0.2747 (2)	0.2140 (3)	0.44349 (17)	0.0595 (8)
C22	0.2713 (2)	0.0741 (3)	0.44226 (19)	0.0632 (8)
H22A	0.2592	0.0261	0.4890	0.076*
C23	0.2853 (3)	0.0031 (4)	0.3737 (2)	0.0784 (10)
H23A	0.2820	-0.0934	0.3725	0.094*
C24	0.3038 (3)	0.0743 (4)	0.3080 (2)	0.0821 (11)
C25	0.3089 (3)	0.2124 (4)	0.3068 (2)	0.0882 (11)
H25A	0.3221	0.2593	0.2600	0.106*
C26	0.2945 (3)	0.2819 (4)	0.3749 (2)	0.0776 (10)
H26A	0.2980	0.3784	0.3754	0.093*
N1	0.21798 (16)	0.3485 (2)	0.78690 (14)	0.0535 (6)
H1	0.2207	0.4340	0.7719	0.066 (9)*
O1	0.26887 (16)	0.13615 (19)	0.76232 (13)	0.0648 (6)
O2	0.2615 (3)	0.4226 (3)	0.50929 (17)	0.1281 (13)
O3	0.4304 (2)	0.2291 (3)	0.62480 (19)	0.1088 (11)
F1	0.3192 (2)	0.0052 (3)	0.24109 (15)	0.1311 (11)

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.107 (3)	0.106 (3)	0.087 (3)	-0.021 (3)	0.061 (3)	-0.016 (3)
C2	0.066 (2)	0.147 (4)	0.092 (3)	0.020 (2)	0.032 (2)	-0.018 (3)
C3	0.069 (2)	0.094 (3)	0.071 (2)	0.0213 (19)	0.0209 (17)	-0.0039 (19)
C4	0.0572 (17)	0.0468 (15)	0.0567 (17)	-0.0004 (13)	0.0220 (13)	-0.0059 (13)
C5	0.106 (3)	0.0581 (19)	0.075 (2)	0.0208 (18)	0.045 (2)	0.0083 (17)
C6	0.160 (4)	0.064 (2)	0.080 (3)	0.010 (2)	0.066 (3)	0.0072 (19)
C7	0.0563 (16)	0.0431 (16)	0.0535 (16)	-0.0041 (12)	0.0186 (13)	-0.0047 (12)
C8	0.0694 (19)	0.0447 (15)	0.0550 (17)	-0.0097 (13)	0.0256 (14)	-0.0043 (13)
C9	0.089 (2)	0.0455 (16)	0.0534 (17)	-0.0118 (15)	0.0210 (15)	-0.0043 (13)
C10	0.086 (2)	0.080 (2)	0.0550 (19)	-0.017 (2)	0.0062 (16)	0.0002 (16)
C11	0.095 (3)	0.098 (3)	0.139 (4)	0.002 (3)	0.001 (3)	0.005 (3)
C12	0.087 (4)	0.169 (6)	0.163 (6)	0.025 (4)	0.003 (3)	0.013 (4)
C13	0.090 (4)	0.215 (8)	0.125 (5)	-0.029 (5)	0.001 (3)	0.002 (5)
C14	0.107 (4)	0.173 (6)	0.113 (4)	-0.059 (4)	0.015 (3)	0.000 (4)
C15	0.103 (3)	0.101 (3)	0.077 (2)	-0.038 (2)	0.018 (2)	-0.002 (2)
C16	0.077 (2)	0.067 (2)	0.073 (2)	-0.0159 (17)	0.0366 (18)	-0.0126 (17)
C17	0.098 (2)	0.116 (2)	0.111 (2)	-0.0421 (18)	0.0240 (18)	-0.0169 (19)
C18	0.165 (6)	0.151 (6)	0.163 (6)	0.002 (4)	-0.051 (5)	0.006 (4)
C19	0.098 (2)	0.116 (2)	0.111 (2)	-0.0421 (18)	0.0240 (18)	-0.0169 (19)
C20	0.117 (3)	0.0525 (19)	0.061 (2)	-0.0195 (18)	0.0282 (19)	-0.0035 (15)

C21	0.070 (2)	0.0604 (19)	0.0477 (16)	-0.0125 (15)	0.0091 (14)	-0.0014 (14)
C22	0.071 (2)	0.061 (2)	0.0578 (18)	-0.0119 (15)	0.0107 (15)	-0.0049 (15)
C23	0.096 (3)	0.068 (2)	0.072 (2)	-0.0158 (19)	0.0166 (19)	-0.0157 (18)
C24	0.097 (3)	0.096 (3)	0.055 (2)	-0.012 (2)	0.0169 (18)	-0.025 (2)
C25	0.121 (3)	0.095 (3)	0.051 (2)	-0.015 (2)	0.0210 (19)	-0.0007 (19)
C26	0.112 (3)	0.068 (2)	0.0532 (19)	-0.0108 (19)	0.0178 (18)	0.0008 (16)
N1	0.0659 (15)	0.0405 (13)	0.0598 (14)	0.0020 (11)	0.0268 (12)	0.0001 (11)
O1	0.0940 (16)	0.0417 (12)	0.0663 (13)	0.0069 (10)	0.0355 (11)	0.0004 (9)
O2	0.261 (4)	0.0547 (16)	0.0844 (19)	-0.0303 (19)	0.076 (2)	-0.0051 (13)
O3	0.0924 (19)	0.129 (2)	0.121 (2)	-0.0246 (17)	0.0629 (18)	-0.0555 (19)
F1	0.195 (3)	0.128 (2)	0.0799 (16)	-0.0225 (19)	0.0503 (17)	-0.0418 (15)

*Geometric parameters (Å, °)*

C1—C2	1.357 (6)	C14—C15	1.376 (7)
C1—C6	1.363 (6)	C14—H14A	0.9500
C1—H1A	0.9500	C15—H15A	0.9500
C2—C3	1.370 (5)	C16—O3	1.201 (4)
C2—H2A	0.9500	C16—C17	1.507 (6)
C3—C4	1.387 (4)	C17—C19	1.499 (6)
C3—H3A	0.9500	C17—C18	1.509 (7)
C4—C5	1.366 (4)	C17—H17A	1.0000
C4—N1	1.419 (3)	C18—H18A	0.9800
C5—C6	1.383 (5)	C18—H18B	0.9800
C5—H5A	0.9500	C18—H18C	0.9800
C6—H6A	0.9500	C19—H19A	0.9800
C7—O1	1.227 (3)	C19—H19B	0.9800
C7—N1	1.341 (3)	C19—H19C	0.9800
C7—C8	1.521 (4)	C20—O2	1.220 (4)
C8—C16	1.516 (5)	C20—C21	1.490 (4)
C8—C9	1.539 (4)	C21—C22	1.375 (4)
C8—H8A	1.0000	C21—C26	1.390 (4)
C9—C10	1.517 (5)	C22—C23	1.379 (4)
C9—C20	1.520 (4)	C22—H22A	0.9500
C9—H9A	1.0000	C23—C24	1.361 (5)
C10—C15	1.372 (5)	C23—H23A	0.9500
C10—C11	1.394 (6)	C24—F1	1.352 (4)
C11—C12	1.387 (7)	C24—C25	1.360 (6)
C11—H11A	0.9500	C25—C26	1.368 (5)
C12—C13	1.357 (8)	C25—H25A	0.9500
C12—H12A	0.9500	C26—H26A	0.9500
C13—C14	1.377 (8)	N1—H1	0.8800
C13—H13A	0.9500		
C2—C1—C6	120.0 (3)	C10—C15—C14	121.7 (5)
C2—C1—H1A	120.0	C10—C15—H15A	119.2
C6—C1—H1A	120.0	C14—C15—H15A	119.2
C1—C2—C3	120.7 (4)	O3—C16—C17	121.6 (4)
C1—C2—H2A	119.7	O3—C16—C8	120.4 (3)
C3—C2—H2A	119.7	C17—C16—C8	118.0 (3)

## supplementary materials

C2—C3—C4	119.7 (4)	C19—C17—C16	110.4 (4)
C2—C3—H3A	120.2	C19—C17—C18	112.4 (5)
C4—C3—H3A	120.2	C16—C17—C18	109.8 (4)
C5—C4—C3	119.6 (3)	C19—C17—H17A	108.0
C5—C4—N1	123.7 (3)	C16—C17—H17A	108.0
C3—C4—N1	116.7 (3)	C18—C17—H17A	108.0
C4—C5—C6	119.8 (3)	C17—C18—H18A	109.5
C4—C5—H5A	120.1	C17—C18—H18B	109.5
C6—C5—H5A	120.1	H18A—C18—H18B	109.5
C1—C6—C5	120.3 (4)	C17—C18—H18C	109.5
C1—C6—H6A	119.9	H18A—C18—H18C	109.5
C5—C6—H6A	119.9	H18B—C18—H18C	109.5
O1—C7—N1	124.0 (2)	C17—C19—H19A	109.5
O1—C7—C8	121.0 (2)	C17—C19—H19B	109.5
N1—C7—C8	115.0 (2)	H19A—C19—H19B	109.5
C16—C8—C7	110.1 (3)	C17—C19—H19C	109.5
C16—C8—C9	111.7 (2)	H19A—C19—H19C	109.5
C7—C8—C9	109.5 (2)	H19B—C19—H19C	109.5
C16—C8—H8A	108.5	O2—C20—C21	119.7 (3)
C7—C8—H8A	108.5	O2—C20—C9	118.3 (3)
C9—C8—H8A	108.5	C21—C20—C9	121.9 (3)
C10—C9—C20	108.3 (3)	C22—C21—C26	118.6 (3)
C10—C9—C8	112.2 (3)	C22—C21—C20	124.2 (3)
C20—C9—C8	110.0 (2)	C26—C21—C20	117.2 (3)
C10—C9—H9A	108.8	C21—C22—C23	120.6 (3)
C20—C9—H9A	108.8	C21—C22—H22A	119.7
C8—C9—H9A	108.8	C23—C22—H22A	119.7
C15—C10—C11	117.3 (4)	C24—C23—C22	118.5 (3)
C15—C10—C9	121.0 (4)	C24—C23—H23A	120.7
C11—C10—C9	121.7 (3)	C22—C23—H23A	120.7
C12—C11—C10	120.5 (5)	F1—C24—C25	118.3 (3)
C12—C11—H11A	119.7	F1—C24—C23	118.8 (4)
C10—C11—H11A	119.7	C25—C24—C23	122.9 (3)
C13—C12—C11	121.2 (6)	C24—C25—C26	118.1 (3)
C13—C12—H12A	119.4	C24—C25—H25A	121.0
C11—C12—H12A	119.4	C26—C25—H25A	121.0
C12—C13—C14	118.7 (6)	C25—C26—C21	121.3 (3)
C12—C13—H13A	120.7	C25—C26—H26A	119.4
C14—C13—H13A	120.7	C21—C26—H26A	119.4
C15—C14—C13	120.6 (6)	C7—N1—C4	126.8 (2)
C15—C14—H14A	119.7	C7—N1—H1	116.6
C13—C14—H14A	119.7	C4—N1—H1	116.6

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 $\cdots$ O1 <sup>i</sup>	0.88	2.10	2.959 (2)	178 (4)
C1—H1A $\cdots$ O3 <sup>ii</sup>	0.95	2.43	3.371 (5)	169
C5—H5A $\cdots$ O1	0.95	2.52	2.963 (4)	109



C26—H26A···O2

0.95

2.41

2.735 (4)

100

Symmetry codes: (i)  $-x+1/2, y+1/2, -z+3/2$ ; (ii)  $x-1/2, -y+1/2, z+1/2$ .

Fig. 1

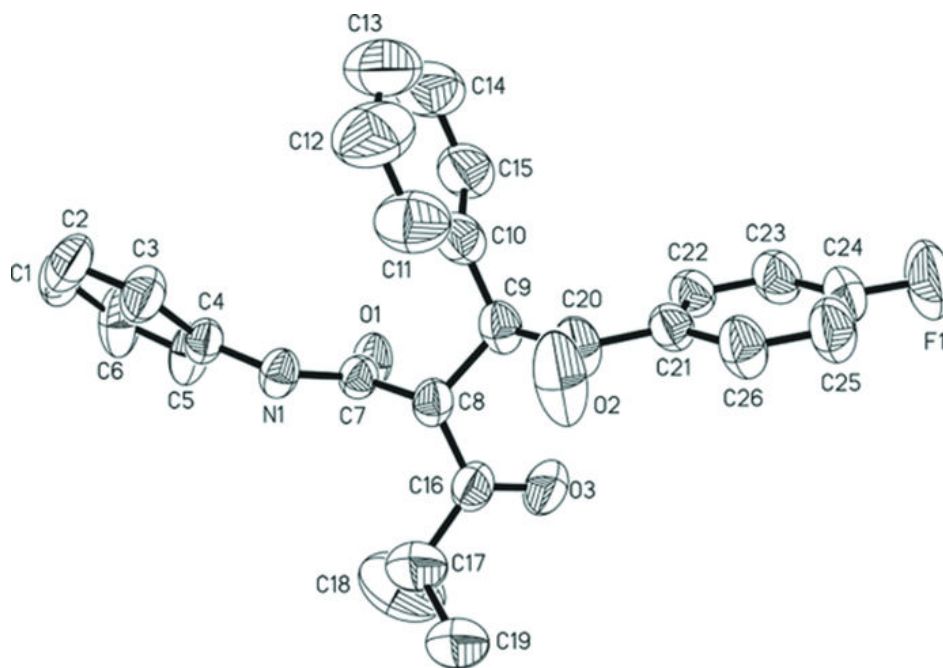


Fig. 2

